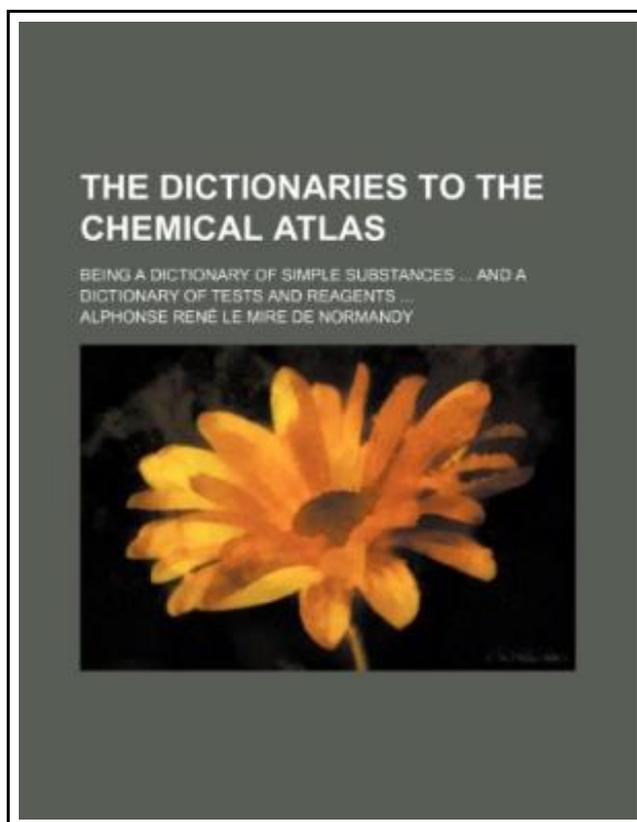


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Rarebooksclub.com, United States, 2012. Paperback. Book Condition: New. 246 x 189 mm. Language: English . Brand New Book ***** Print on Demand *****.This historic book may have numerous typos and missing text. Purchasers can download a free scanned copy of the original book (without typos) from the publisher. Not indexed. Not illustrated. 1857 edition. Excerpt: .it the cold mixture of sulphuric acid and water above alluded to. Connect the retort with a refrigerator or Liebig s condenser, and distil the whole to dryness by means of a sand-bath. Or else reduce to powder 3 parts of dry acetate of soda, pour upon it 9-7 parts of pure and very concentrated SO₃, HO, and at once connect the whole with a Liebig s condenser; one-eighth part of the acetic acid will spontaneously distil over by the heat developed by the mixture itself, and the rest of the acetic acid will pass over upon applying a gentle heat to the mixture. The distillate will probably require purification, which is accomplished by submitting it to a second distillation. If it has been found that it is contaminated by SO_a, the best way will be to digest in it some pure oxyde of lead (PbO₂), which will transform it into SO₃, and combine with it; the superincumbent liquor may then be carefully decanted, and further purified by redistillation. Acetic acid is sometimes used for acidifying liquors when the use of mineral acids. would be objectionable, or to precipitate by HS certain substances, oxyde of zinc for example, the solutions of which in the mineral acids cannot be precipitated by that reagent. The acetic acid fortis of commerce is good enough for such a purpose. Acetic acid is used also to separate phosphate from oxalate of lime; the former being soluble, and the latter insoluble...

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